



Biosynthesis of silver nanoparticles from *Aeromonas sobria* and antibacterial activity against fish pathogens

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Abstract

Fish diseases are a major obstacle to the development of the fisheries industry. *Aeromonas sobria* is an infectious waterborne bacterium that causes ulcers, tail rot and hemorrhagic septicemia in fishes and resistant to many existing antibiotics. In this context, *A. sobria*-AgNPs were synthesized by *A. sobria* using AgNO₃. *A. sobria*-AgNPs were characterized using UV–Vis spectroscopy, and a peak was obtained at a range of 420–480 nm. *A. sobria*-AgNPs were evaluated for antibacterial activities against different fish pathogens. The highest antibacterial activity was observed against *A. hydrophila*, *E. cloacae* and *E. coli*. The lower activity was found against *C. braakii* and *E. hermannii*, but against *H. alvei*, *P. rettger* and *M. organii* subsp. *sibonii* no zone of inhibition was recorded. The results indicated that the *A. sobria*-AgNPs can be used to develop antibacterial agent and as a therapeutic agent in the fishing industry and water disinfection. The antibacterial efficacy against the fish pathogen *A. hydrophila* of silver nanoparticles is a hope for possible application as a disinfectant or antimicrobial agent for better fish health management.

Keywords *Aeromonas sobria* · Antibacterial activity · Silver nanoparticles

Introduction

Recently, there has been an increase in the fishing industry. Fish is a healthy food because it is rich in quality protein and fatty acids (Tacon and Metian 2013). For this reason, fish can provide an important food source for humans. In addition, it also provides added value to livelihoods, job creation, income generation and, most importantly, global food security (Gustafson 2013). With appropriate policies, it is necessary to ensure that both fish farming and aquaculture production are sustainable. Therefore, governments need to manage ecological and social impacts in aquaculture. In this sector, in order to grow healthy fish, it must first be purified from pathogenic microorganisms. Fish diseases are the cause of great economic losses in fish

farming. The global economic impact of diseases causes serious economic losses in finfish aquaculture estimated between 0.16 and 4.0 billion USD per year (Kubitza 2005; Martins et al. 2008). *Aeromonas* are isolated from food, clinical and environmental samples and fish (Beaz-Hidalgo et al. 2013). In this environment, *A. sobria* virulent strains are a possible source of infection (Janda and Abbott 2010; Rathore et al. 2005). In order to grow and feed healthy fish, it is necessary to remove fish pathogens such as *Aeromonas* from the environment where fish lives. Since bacteria have developed resistance against antibiotics in recent years, alternative new treatment methods are needed. For this reason, nanotechnology research has gained worldwide momentum in recent years (Glisovic et al. 2017). Metal nanoparticles are increasingly being used for biological and environmental safety. Recently, noble metal nanoparticles (Ag, Zn, Mn) have become the focus of attention with their new applications in biotechnology, catalysis, electronics, environment and optics (Dastafkan et al. 2015; Ahadi et al. 2016). Although chemical and physical approaches from basic methods are used for nanoparticle production, plant extracts and microorganisms have been actively pursued in recent years as an alternative method (Singh and Prasad 2017). Nanoscience and nanotechnology, especially

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nanomedicine, silver–bacteria interaction, have found application areas. AgNPs can be used as antifungal, antiviral, anti-inflammatory, anti-angiogenic and anticancer agents as a result of biosynthesis with microorganisms (Zhang et al. 2016). They are commonly used in diagnostic and therapeutic products, wound dressings, air and water treatment, paint and food packaging (Li et al. 2014). In addition, silver nanoparticles (AgNPs) are used as alternative antimicrobial agents in antibiotics for the treatment of infections caused by highly resistant bacteria in humans, fish and other animals (Akram et al. 2006). Therefore, antibacterial characteristics of nanoparticle synthesized by environmentally friendly microorganisms known as small nanofactories have been used against the bacterial fish diseases in aquaculture (Raut et al. 2009). Due to the increasing demand for nanoparticles in treatment of fish pathogens, several studies are carried out for the synthesis of nanoparticles from different microorganisms (Whitesides 2003; Fortin and Beveridge 2000). Extracellular biosynthesis of AgNPs from *E. coli*, *S. aureus*, *P. aeruginosa*, *Morganella* sp. and *A. hydrophila* was carried out by several researchers (Gurunathan et al. 2009). There is no study of silver nanoparticle synthesis from *A. sobria*, which is the cause of disease in fishes such as *A. hydrophila*.

The aim of this work was to study the synthesis and characterization of AgNPs using *A. sobria* isolated from organs of fish of Sıdıklı Küçükboğaz Dam Lake. Their antibacterial activity was also evaluated against isolated fish pathogens.

Materials and methods

Isolation and identification of bacterial strain

The bacteria in this study were isolated from the organs of *Cyprinus carpio* L. and *Tinca tinca* L. collected from Sıdıklı Küçükboğaz Dam Lake. Nutrient agar was plated with fish organs (gill, intestine) and incubated at 37 °C for 72 h. Bacteria were identified using the VITEK 2 system. Thirty-seven isolated bacteria were purified and identified

at species level. *A. sobria* was selected from the bacteria isolated from fishes and investigated in this study.

Synthesis of AgNPs

For AgNPs synthesis, the culture *A. sobria* was freshly inoculated into nutrient broth and incubated for 72 h on shaker at 120 rpm at 30 °C. The culture of *A. sobria* was centrifuged at 12,000 rpm for 10 min, and the supernatant was used for the synthesis of AgNPs. The four test tubes, the first containing AgNO₃ (Sigma, USA, 99.9% purity) without the supernatant, the second containing only the media and the third and fourth containing the supernatant and AgNO₃ solution in 250 ml aliquots at concentrations of 0.5 mM and 1 mM, respectively, were incubated at 30 °C for 18, 24 and 72 h. The absorption spectrum of the sample was recorded on the range of 420–480 nm using a UV-1800 spectrophotometer (Kumar and Mamidara 2011).

UV–Visible spectroscopic analysis

UV–Vis spectrum analysis was performed in the range of 420–480 nm using a UV-1800 spectrophotometer. The reduction of Ag⁺ ions was measured at different time intervals: 18, 24 and 72 h. AgNPs were visually examined by the color change in the culture medium from light brown to brown, and nanoparticles were confirmed by measuring AgNPs with UV–vis spectrum.

Determination of antibacterial activity

Disk diffusion test

The antibacterial activity of the synthesized AgNPs was assessed by well diffusion method against pathogenic organisms such as *Aeromonas veronii*, *Hafnia alvei*, *Aeromonas hydrophila*, *Pseudomonas oleovorans*, *Citrobacter braakii*, *Escherichia hermannii*, *Aeromonas sobria*, *Providencia rettgeri*, *Morganella morganii* subsp. *sibonii*, *Enterobacter cloacae* and *Escherichia coli*. Bacterial cultures were also grown



in trypticase soy broth, and each strain was cultivated with individual trypticase soy agar using sterile cotton swabs. Then, wells of 6 mm diameter were made on trypticase soy agar plates using gel puncture. For the biosynthesized silver nanoparticle, sterile distilled water was taken as a control group and other 10 wells were loaded with 75 μL of silver nanoparticles of 5 $\mu\text{g}/\text{ml}$, 2.5 $\mu\text{g}/\text{ml}$, 1.25 $\mu\text{g}/\text{ml}$, 0.75 $\mu\text{g}/\text{ml}$, 0.37 $\mu\text{g}/\text{ml}$, 0.187 $\mu\text{g}/\text{ml}$, 0.09 $\mu\text{g}/\text{ml}$, 0.046 $\mu\text{g}/\text{ml}$, 0.023 $\mu\text{g}/\text{ml}$ and 0.01 $\mu\text{g}/\text{ml}$ concentrations of original nanoparticles solution. After incubation at 37 $^{\circ}\text{C}$ for 48 h, diameter (in mm) of the obvious clear zones was measured.

Minimum inhibitory concentration (MIC) determination

A broth microdilution method was used to determine the toxicity of AgNPs to pathogenic bacteria. Similar concentrations of AgNPs showed antimicrobial activity by agar well diffusion method. Bacterial growth was assessed in the presence and absence of AgNPs.

Statistical analysis

Agar well diffusion and MIC were performed in triplicate, and the results were expressed as means \pm the standard deviation of the means. SPSS academic software was used for statistical calculations. P values lower than 0.05 were considered significant.

Results and discussion

Synthesis of AgNPs

When the *A. sobria* is exposed to AgNO_3 , the color turns into dark brown within a few minutes after the reaction started, indicating the formation of AgNPs. The formation of the dark brown color intensity with the incubation time of the reaction was directly proportional. The concrete increase in absorbance with color intensity was revealed by periods of time of up to 72 h, as shown in Fig. 1. The formation of brown color at 30 min in the solution (bacteria with silver nitrate solution) reveals the reduction of silver ions into silver nanoparticles. The maximum color intensity was attained after 24 h. The color change was exhibited due to the oscillation of electron in the silver nanoparticles (Duran et al. 2007).

UV-visible spectroscopic analysis

The AgNPs were characterized by UV-Vis spectrophotometry. The reaction in *A. sobria* was determined at different times (18, 24 and 72 h). The UV-Vis spectrum is graphically shown in Fig. 1. In the biosynthesis of AgNPs, the nanoparticle production was monitored by color change and then optically measured by UV-Vis spectrophotometer; this analysis showed an absorbance peak shifted from 480 to 420 nm. This band shift shows that the synthesis process of the particles increases when the incubation time increases (Fig. 1). This increase in absorbance due to color intensity can be attributed to the increase in the number of

Fig. 1 The UV-visible spectra of AgNPs prepared by the supernatant of *A. sobria* (after 18-, 24- and 72-h incubation)

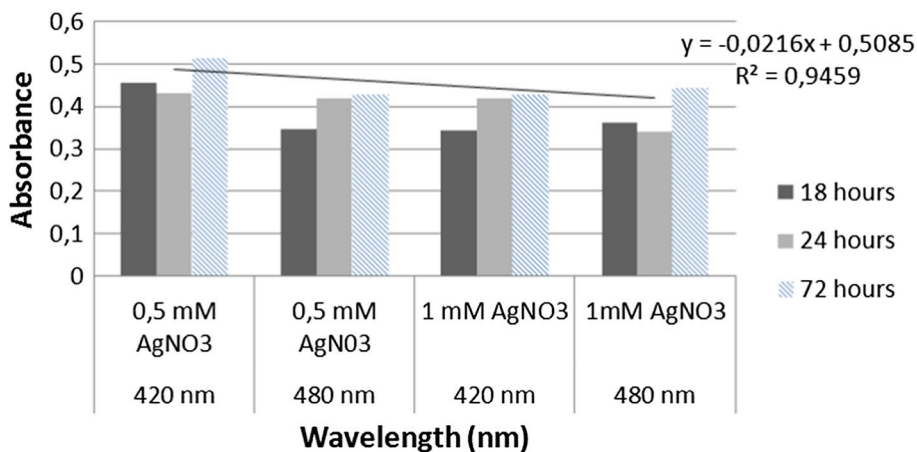


Table 1 The diameter of inhibition zone (DIZ) and minimum inhibitory concentration (MIC) of AgNPs (5 µg/mL) synthesized by *A. sobria*

Indicator strains	Zone of inhibition (mm) ^c	^a MIC (µg/ml) ^d
<i>Aeromonas veronii</i>	12 (± 1)	0.37 (± 1)
<i>Hafnia alvei</i>	NCD ^c	NCD ^c
<i>Aeromonas hydrophila</i>	10 (± 1)	0.187 (± 0.1)
<i>Pseudomonas oleovorans</i>	12 (± 1)	0.37 (± 0.05774)
<i>Citrobacter braakii</i>	2 (± 1)	0.37 (± 1)
<i>Escherichia hermannii</i>	8 (± 1)	0.37 (± 1)
<i>Aeromonas sobria</i>	12 (± 1)	0.023 (± 0.1)
<i>Providencia rettger</i>	NCD ^c	NCD ^c
<i>Morganella morganii</i> subsp. <i>sibonii</i>	NCD ^c	NCD ^c
<i>Enterobacter cloacae</i>	12 (± 1)	0.187 (± 0.1)
<i>Escherichia coli</i>	15 (± 1)	0.37 (± 1)
Negative control ^b	0	0

^aMIC minimum inhibitory concentration

^bNegative control: flask containing inoculum and peptone water medium, devoid of nanoparticles

^cNCD no culturable cells detected

^dMean ± standard deviation of the mean

silver nanoparticles over time (Bhainsa and D'Souza 2006). In the biosynthesis of metal nanoparticles, metal ions are completely reduced by bacteria and fungi for 24–124 h (Korbekandi et al. 2009). Together with the intensity of color, the increase in absorbance was revealed with time periods of 72 h. Our work is comparable to other studies.

Antibacterial assay

The effect of AgNPs on bacteria was performed using disk diffusion method. Biologically synthesized silver

nanoparticles showed great antibacterial activity against *A. sobria*. The biosynthesized nanoparticles showed antimicrobial activity against most bacteria at concentrations of 0.37 µg/mL, 0.187 µg/mL and 0.023 µg/mL.

It is observed that the synthesized silver nanoparticles exhibited antibacterial activity against *A. sobria* by both the well diffusion method and minimal inhibitory concentration (MIC) (Table 1).

The antimicrobial activity of AgNPs was investigated against various bacteria such as *A. veronii*, *H. alvei*, *A. hydrophila*, *P. oleovorans*, *C. braakii*, *E. hermannii*, *A. sobria*, *P. rettger*, *M. morganii* subsp. *sibonii*, *E. cloacae* and *E. coli* using well diffusion method. The highest antibacterial activity was observed against *A. veronii*, *A. hydrophila*, *P. oleovorans*, *A. sobria*, *E. cloacae* and *E. coli*. The lower activity was found against *C. braakii* and *E. hermannii*, but against *H. alvei*, *P. rettger* and *M. morganii* subsp. *sibonii* no zone of inhibition was recorded. The mean of three replicates of the diameter of inhibition zones (in millimeters) around each well with AgNPs solution is represented in Table 1.

Shaffiey et al. (2014) reported that synthesized AgNPs exhibited moderate activity toward *A. hydrophila*. In one study, AgNPs were reported to exhibit good antibacterial activity against *E. coli* and *B. subtilis* (Anandalakshmi et al. 2016). Because phyto-synthesized AgNPs are entirely of natural origin, they can be administered as an alternative to antibiotics and biocides and therapeutic agents against *A. hydrophila* induced diseases in aquatic animals (Mahanty et al. 2013). The antibacterial efficacy against the fish pathogen *A. hydrophila* of silver nanoparticles is a hope for possible application as a disinfectant or antimicrobial agent for better fish health management (Sarkar et al. 2012). All test bacteria (about 10⁸ cells/mL) were produced for 24 h at AgNPs concentrations (5 µg/mL). The MIC values of AgNPs are presented in Table 1. The results show the effectiveness of AgNPs against *A. sobria* with low MIC



(0.023 µg/ml). The biosynthesized nanoparticles showed antimicrobial activity against most bacteria at concentrations of 0.37 µg/mL, 0.187 µg/mL and 0.023 µg/mL. Similarly, Haytham (2015) examined the MIC of *B. subtilis*, *S. aureus*, *P. aeruginosa* and *E. coli* against the silver nanoparticle at different concentrations (6.8, 5.1, 1.70 and 3.4 mg/ml). For the first time, this study shows the synthesis of silver nanoparticles and antibacterial activity against fish pathogens using *A. sobria*-AgNPs at different time intervals (18 h, 24 h and 72 h) with UV spectrum.

Conclusion

AgNPs presented the best antimicrobial activity test against *A. veronii*, *A. hydrophila*, *P. oleovorans*, *A. sobria*, *E. cloacae* and *E. coli*. AgNPs can be used as an alternative to antibiotics in fisheries and other aquaculture production. For this reason, further in vivo studies are needed.

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