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# Chinese Journal of Physics

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## Investigation of structural, electronic, elastic and phonon properties of cubic spinel  $\text{ZnM}_2\text{O}_4(M = \text{Co}, \text{Rh} \text{ and } \text{Ir})$  compounds

### Mustafa Özduran

Department of Physics, Faculty of Arts and Sciences, Ahi Evran University, 40100, Turkey

#### ARTICLE INFO

Keywords: DFT Spinel oxides Electronic band structure Phonon dispersion Elastic constants

#### ABSTRACT

The results obtained from ab initio calculations on  $\text{ZnM}_2\text{O}_4$  (M = Co, Rh and Ir) compounds have been reported. The elastic constants, Bulk, Shear and Young modulus, and Poisson's ratios of the compounds are presented. In addition, full phonon dispersion curves and projected density of states of the compounds have been computed using the direct method. The lattice parameters (a) and internal parameters  $(u)$  are found to be in a good agreement with experimental results. According to both the  $B/G$  values and the Poisson's ratio, these compounds have covalent bondings. The analysis of the band structure of these compounds have indicated indirect band gaps of 1.25 eV for  $ZnCo<sub>2</sub>O<sub>4</sub>$  and 1.14 eV for  $ZnRh<sub>2</sub>O<sub>4</sub>$  and 0.86 eV for  $ZnIr<sub>2</sub>O<sub>4</sub>$ . The full phonon spectra of these compounds show that they are dynamically stable in the cubic spinel structure.

#### 1. Introduction

Transparent conductive oxides (TCOs) are electrically conductive materials in the visible region of the spectrum and have very low absorption of electromagnetic waves. These materials are usually synthesized using thin film technology and utilised in optoelectric circuit components such as solar cells and image screens. In addition, they can be manufactured as n-type or p-type conductive materials and have applications in opto-electronic technology which in turn provides saving of power in wide ranges of applications [\[1,2\]](#page-7-0). Other synthesis methods include scattering, reactive thermal deposition, chemical vapor deposition, laser ablation, and sol-gel method [3–[7\]](#page-7-1).

Zinc spinel ZnM<sub>2</sub>O<sub>4</sub> (M = Co, Rh, Ir) compounds have been a great of interest in many experimental and theoretical works  $[8-29]$  $[8-29]$ . The structural and electronic properties of ZnM<sub>2</sub>O<sub>4</sub> (M = Co, Rh, Ir) spinel compounds were calculated by Amini et al. using Heyd-Scuseria and Ernzerhof (HSE06) functions [\[8\].](#page-7-2) In their study, they found that the indirect band gap values for  $ZnCo<sub>2</sub>O<sub>4</sub>$ ,  $\text{ZnRh}_2\text{O}_4$  and  $\text{ZnIr}_2\text{O}_4$  compounds as 3.88, 2.91 and 2.53 eV, respectively. Volnianska et al. [\[9\]](#page-7-3) examined the structural and electronic properties of spinel ZnRh<sub>2</sub>O<sub>4</sub> compound using the density functional theory in the generalized gradient approach (GGA) with Quantum-Espresso code and reported a lattice constant of 8.67 Å for ZnRh<sub>2</sub>O<sub>4</sub>. Ramo et al. [\[11\]](#page-7-4) investigated the influence of imperfections and irregularities on the electronic properties of  $\rm Zn_2O_4$  with the CRYSTAL09 package and the hybrid Becke threeparameter Lee–Yang–Parr (B3LYP) function. Dekkers et al. [\[13\]](#page-7-5) measured the lattice constants of the p-type transparent conductive  $ZnCo<sub>2</sub>O<sub>4</sub>$ ,  $ZnRh<sub>2</sub>O<sub>4</sub>$  and  $ZnIr<sub>2</sub>O<sub>4</sub>$  oxides as 8.104 Å, 8.489 Å and 8.507 Å, respectively, using the X-ray diffraction method. In the same study, they measured the band gap values of  $ZnCo<sub>2</sub>O<sub>4</sub>$ ,  $ZnRh<sub>2</sub>O<sub>4</sub>$  and  $ZnIr<sub>2</sub>O<sub>4</sub>$  as  $2,26 \text{ eV}$ ,  $2,74 \text{ eV}$  and  $2,97 \text{ eV}$ , respectively, by vibrating laser storage method. The lattice constant of spinel  $ZnCo<sub>2</sub>O<sub>4</sub>$  was measured by XRD method and found to be 8.120 Å in an experimental study conducted by Sharma et al. [\[14\].](#page-8-0) The optical properties and electronic structure of  $\text{ZnRh}_2\text{O}_4$  spinel have been investigated in an experimental study by Singh et al.  $[15]$  and the indirect band-gap for ZnRh<sub>2</sub>O<sub>4</sub> was measured as 1.2 eV. The

<https://doi.org/10.1016/j.cjph.2019.02.024>

Received 15 August 2018; Received in revised form 15 January 2019; Accepted 20 February 2019 Available online 28 February 2019

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E-mail address: [mozduran@ahievran.edu.tr](mailto:mozduran@ahievran.edu.tr).

electronic and optical properties of  $\text{ZnM}_2\text{O}_4$  (M = Co, Rh, Ir) spinel oxides have recently been studied by Samanta [\[25\]](#page-8-2) using both PBE and Tran-Blaha modified Becke-Johnson density function. He found indirect band gap value for  $ZnCo<sub>2</sub>O<sub>4</sub>$  as 0.773 eV with PBE and 3.354 eV with TB-mBJ function, for ZnRh<sub>2</sub>O<sub>4</sub> as 0.897 eV with PBE and 2.53 eV with TB-mBJ function, for ZnIr<sub>2</sub>O<sub>4</sub> as 0.416 eV with PBE and 2.32 eV with TB-mBJ function, respectively. Those studies in literature however do not present a full thorough investigation on these materials, for instance, the phonon spectrum of the solids are very important in determining various basic solid state properties such as thermal conductivity, thermal expansion, heat conduction, electron-phonon interaction, specific heat, and phase transition. The full phonon properties of  $ZnCo<sub>2</sub>O<sub>4</sub>$ ,  $ZnRh<sub>2</sub>O<sub>4</sub>$  and  $ZnIr<sub>2</sub>O<sub>4</sub>$  have not yet been studied using any theoretical or experimental method to best of our knowledge Thus, in this study, the phonon properties as well as the structural, elastic and electronic properties of ZnCo<sub>2</sub>O<sub>4</sub>, ZnRh<sub>2</sub>O<sub>4</sub> and ZnIr<sub>2</sub>O<sub>4</sub> spinel compounds were investigated using the ab initio calculations of density functional theory (DFT). This work specifically focuses on the study of the full phonon spectrum of  $ZnCo<sub>2</sub>O<sub>4</sub>$ ,  $ZnRh<sub>2</sub>O<sub>4</sub>$  and  $ZnIr<sub>2</sub>O<sub>4</sub>$  spinel compounds using the direct method.

### 2. Method

Ab initio calculations were performed using plane wave pseudopotential density functional theory (DFT) as implemented in the code MedeA [\[30,31\]](#page-8-3) with the projected augmented wave [\[32\]](#page-8-4). The generalized gradient approximation (GGA) of Perdew-Burke–Ernzerhof (PBE) [\[33\]](#page-8-5) was chosen for exchange-correlation functional. An energy cutoff of 500 eV was found in the calculation of structural, elastic and phonon properties. Energy convergence criteria were set to  $10^{-9}$  eV using the normal (blocked Davidson) algorithm and reciprocal space projection operators. The Brillouin was integrated using Monkhorst–Pack generated sets of k-points [\[34\].](#page-8-6) For the ZnRh<sub>2</sub>O<sub>4</sub>, ZnIr<sub>2</sub>O and ZnCo<sub>2</sub>O<sub>4</sub> spinellars,  $5 \times 5 \times 5$  and  $6 \times 6 \times 6$  k-points meshes were found to be sufficient, respectively. Fermi level was smeared by Methfessel–Paxton type [\[35\]](#page-8-7) with a width of 0.225 eV. Elastic constants were calculated using the stress-strain approximation [\[36\],](#page-8-8) where stress due to an applied strain was directly calculated. Phonon spectra were obtained using the MedeA-Phonon in conjunction with MedeA-Vasp by the direct method.  $1 \times 1 \times 1$  supercell containing 56 atoms for the spinel structure were used. The calculated specific heat capacity at constant volume  $(C_v)$  of zinc spinel ZnM<sub>2</sub>O<sub>4</sub> (M = Co, Rh, Ir) compounds were calculated using the quasi harmonic approximation (QHA).

#### 3. Results

#### 3.1. Structural properties

 $\text{ZnM}_2\text{O}_4$  (M = Co, Rh, Ir) are spinel oxides compounds, which belong to Fd-3m (227) space group. Spinel compounds are characterized by two structural constants. The first one is the lattice constant  $(a_0)$ , the second is the internal parameter  $(u)$ . The Zn atoms are located at tetrahedral positions 8a (1/8, 1/8, 1/8) at Wyckoff positions, while the M (Co, Rh, Ir) atoms are located at 16d  $(1/2, 1/2, 1/2)$  and O atoms are surface-centered cubic at 32e (u, u, u) [\[37\].](#page-8-9) The crystal structure of cubic spinel oxides  $\text{ZnM}_2\text{O}_4$ ( $M = Co$ , Rh, Ir) is shown in [Fig. 1](#page-1-0). For spinel  $ZnM_2O_4$  ( $M = Co$ , Rh, 1r) compounds, crystal structures were formed using the InfoMaticA Database included in the MedeA package program. Then the convergence value of the total energy for k points and the cut-off energy E<sub>cut</sub> were found. The energies of these materials were 532 eV for  $ZnCo<sub>2</sub>O<sub>4</sub>$ ,  $ZnRh<sub>2</sub>O<sub>4</sub>$  compounds, 510 eV for  $ZnIr<sub>2</sub>O<sub>4</sub>$ and the values of k-points were taken as  $6 \times 6 \times 6$  ZnRh<sub>2</sub>O<sub>4</sub> for ZnCo<sub>2</sub>O<sub>4</sub> and  $5 \times 5 \times 5$  for ZnIr<sub>2</sub>O<sub>4</sub> compounds. The computed the

<span id="page-1-0"></span>

Fig. 1. The crystal structure of cubic spinel  $\text{ZnM}_2\text{O}_4$  (M = Co, Rh, Ir) compounds.

#### <span id="page-2-0"></span>Table 1

The computed lattice constants a (Å), internal structure parameters u and Bulk modulus B(GPa), Debye temperature  $\Theta_D$  (K), elastic costants C<sub>ij</sub> (GPa), Shear modulus G(GPa), B/G ratio, Young modulus E(GPa) and Poisson ratio  $\nu$  of ZnM<sub>2</sub>O<sub>4</sub> (M = Co, Rh, Ir) compounds.

Compounds	References	$\mathbf{a}$	$\mathbf u$	B	$\Theta_{\rm D}$	$C_{11}$	$C_{12}$	$\rm{C}_{44}$	${\bf G}$	B/G	E	ν
ZnCo <sub>2</sub> O <sub>4</sub>	This work	8.159	0.2649	175.21	505,616	253.69	135.97	96.87	79.33	2.21	206.77	0.303
	[8]	8.02	0.264	$\overline{\phantom{m}}$	$\overline{\phantom{0}}$	$\overline{\phantom{0}}$	$\overline{\phantom{0}}$	-	-	-	-	
	$[13-Exp.]$	8.104	-	$\overline{\phantom{0}}$	-	$\overline{\phantom{0}}$	-	-	$\overline{\phantom{0}}$	-	۰	-
	$[14]$	8.120	-	-	-	-	-	-	-	-	-	
	[20]	8.0573	$\overline{\phantom{0}}$	$\overline{\phantom{0}}$	÷	$\overline{\phantom{0}}$	÷,	-	$\overline{\phantom{0}}$	÷,	-	-
	$[10]$	8.176	0.264	$\overline{\phantom{0}}$	-	-	-	-	-	-	۰	
	$[24]$	8.104	0.263				-					
	$[25]$	8.176	0.264	210.34	-		-					
	$[26]$	8.05	-	$\overline{\phantom{0}}$	-	-	-			-		
	$[27-PBE]$	8.17	-	$\overline{\phantom{0}}$	-	$\overline{\phantom{0}}$	-	-		-		
	$[27-HSE]$	8.01	÷									
ZnRh <sub>2</sub> O <sub>4</sub>	This work	8.618	0.2599	190.59	384,991	255.86	157.96	61.36	59.05	3.40	153.15	0.366
	[8]	8.49	0.26		-		-		-			
	[9]	8.67	-	$\overline{\phantom{0}}$	-	-	-	-	-	-	-	-
	$[13-Exp.]$	8.489	$\overline{\phantom{0}}$		-		-					
	$[15]$	8.506	0.261		-		-			-	۰	
	$[24]$	8.489	0.263									
	$[25]$	8.637	0.261	198.19	$\overline{\phantom{0}}$	$\overline{\phantom{0}}$	$\overline{\phantom{0}}$		$\overline{\phantom{0}}$	-	۳	-
	$[27-PBE]$	8.57	-	-	-	-	-	-				
	$[27-HSE]$	8.48	-		-							
$\text{ZnIr}_2\text{O}_4$	This work	8.745	0.2598	181.09	281,557	250.56	146.35	44.99	47.71	3.79	131.58	0.379
	[8]	8.59	0.26	$\overline{\phantom{0}}$	-	$\overline{\phantom{0}}$	$\overline{\phantom{0}}$	-	-	-	-	
	[11]	8.77	$\overline{\phantom{0}}$	$\overline{\phantom{0}}$	-	-	-	-	$\overline{\phantom{0}}$	-	-	-
	$[13-Exp.]$	8.507	÷.	-	-	-	-	-	-	-	-	
	$[24]$	8.507	0.263		-		-	-	-	-	-	
	$[25]$	8.759	0.262	207.43	-	-	-	-	-	-	-	
	$[27-PBE]$	8.75	-	$\overline{\phantom{0}}$	-							
	$[27-HSE]$	8.59	$\overline{\phantom{0}}$		÷							

internal parameter (u), lattice constants  $(a_0)$  for  $\text{ZnM}_2\text{O}_4$  (M = Co, Rh, Ir) compounds are given in [Table 1](#page-2-0) and compared with previous experimental and theoretical studies [8–[11, 13](#page-7-2)–15, 20–22, 24–27]. These calculated values are in a good agreement with the previously values. Moreover, the calculated internal parameters for spinel  $\text{ZnM}_2\text{O}_4$  (M = Co, Rh, Ir) compounds were compared with the theoretical work Ref. [\[8\],](#page-7-2) and the error values for  $ZnCo<sub>2</sub>O<sub>4</sub>$ ,  $ZnRh<sub>2</sub>O<sub>4</sub>$  and  $ZnIr<sub>2</sub>O<sub>4</sub>$  spinel compounds were found to be 0.35%, −0.03% and −0.07%, respectively. The calculated lattice constants in this study are compared with Ref. [\[21\],](#page-8-10) the error values of the lattice constants were calculated as 0.68%, 1.63% and 2.79% for spinel  $\text{ZnCo}_2\text{O}_4$ ,  $\text{ZnRh}_2\text{O}_4$  and  $\text{ZnIr}_2\text{O}_4$  compounds, respectively. The computed values of the bulk modulus of spinel oxides compounds are predicted to be 175.21 GPa, 190.59 GPa, and 181.09 GPa for  $ZnCo<sub>2</sub>O<sub>4</sub>$ ,  $ZnRh<sub>2</sub>O<sub>4</sub>$  and  $ZnIr<sub>2</sub>O<sub>4</sub>$  compounds, respectively. These computed values are good agreement with available theoretical data [\[25\].](#page-8-2)

#### 3.2. Electronic properties

The electronic band structures along the high symmetry directions for spinel oxides  $\text{ZnM}_2\text{O}_4$  (M = Co, Rh, Ir) compounds are presented in [Fig. 2](#page-3-0). From the electronic band structure graphs, the direct and indirect band gap values of these compounds are given in [Table 2](#page-4-0). Dekkers et al. [\[13\]](#page-7-5) found that the values obtained in the study using the HSE06 method were in agreement with our work and that the indirect band gaps at Γ–X directions were 2.26 eV, 2.74 eV, 2.97 eV for ZnCo<sub>2</sub>O<sub>4</sub>, ZnRh<sub>2</sub>O<sub>4</sub> and ZnIr<sub>2</sub>O<sub>4</sub> spinel com-pounds, respectively, and also Amini et al. [\[8\]](#page-7-2) used the another method to find the indirect band gaps of  $ZnCo<sub>2</sub>O<sub>4</sub>$ ,  $ZnRh<sub>2</sub>O<sub>4</sub>$  and ZnIr2O4 as 3.88 eV, 2.91 eV and 2.53 eV, respectively. The indirect band gap values for the three compounds are given in [Table 2](#page-4-0) with available data. Moreover, for better understanding of electronic band contribution of these compounds, the total and partial density of state curves of the compounds were obtained. The density of state curves obtained for spinel oxides  $ZnCo<sub>2</sub>O<sub>4</sub>$ ,  $ZnRh<sub>2</sub>O<sub>4</sub>$  and  $ZnIr<sub>2</sub>O<sub>4</sub>$ compounds are shown in [Fig. 3](#page-5-0). It can be seen from the densities of state that for the spinel  $ZnCo_2O_4$ ,  $ZnRh_2O_4$  and  $ZnIr_2O_4$ compounds, the main contributions below and above the band gap for three compounds came from electrons of the 3d and 2p orbitals. The Fermi energies of the compounds were calculated as  $E_F$  (ZnCo<sub>2</sub>O<sub>4</sub>) = 5.28 eV,  $E_F$  (ZnIr<sub>2</sub>O<sub>4</sub>) = 5.03 eV and  $E_F$  $(ZnRh<sub>2</sub>O<sub>4</sub>)$  = 5.55 eV and the Fermi energy was taken as zero by subtracting from all other band energies. The indirect band gap for all three compounds, at Γ–X direction have been found to be different values with different methods.

#### 3.3. Elastic properties

Elastic constants are significant parameters that describes the mechanical behaviour of a crystal and provide information about the strength and the structural stability of the material. There are three independent elastic constants such as  $C_{11}$ ,  $C_{12}$  and  $C_{44}$  for cubic systems. The calculated values for Shear modulus (G), Bulk modulus (B) and Young's modulus (E), elastic anisotropy factors (A),

<span id="page-3-0"></span>

Fig. 2. The computed electronic band structure of cubic spinel  $\text{ZnM}_2\text{O}_4$  (M = Co, Rh, Ir) compounds.

Poisson's ratios and elastic constants  $(C_{ij})$  of spinel oxides ZnM<sub>2</sub>O<sub>4</sub> (M = Co, Rh, Ir) are summarized in [Table 2](#page-4-0). According to the Born stability criteria [\[38\]](#page-8-15), the computed values of the elastic constants is indicative of the mechanical stability of ZnM<sub>2</sub>O<sub>4</sub> (M = Co, Rh, Ir) in the spinel structure.

#### <span id="page-4-0"></span>Table 2

The computed values of direct band gaps  $(\Gamma - \Gamma, L - L, X - X)$  and indirect band gaps  $(\Gamma - L, \Gamma - X)$  for ZnM2O4 (M = Co, Rh, Ir) compounds.

Compounds	References	$\Gamma=\Gamma$	$\mathbf{L}-\mathbf{L}$	$X - X$	$\Gamma$ $-$ L	$\Gamma$ – X
ZnCo <sub>2</sub> O <sub>4</sub>	This work	2.21	1.65	0.94	1.76	1.25
	$[8-HSE]$	$\overline{\phantom{a}}$	$\overline{\phantom{0}}$	$\overline{\phantom{0}}$	÷.	3.88
	$[13-Exp.]$	$\overline{\phantom{a}}$	$\overline{\phantom{a}}$	$\overline{\phantom{0}}$	$\overline{\phantom{0}}$	2.26
	$[16-Exp.]$		-	-	$\overline{\phantom{0}}$	2.63
	$[24 - PBE]$		$\equiv$	$\equiv$		0.71
	$[24-TB-mBJ]$	$\overline{\phantom{0}}$	$\overline{\phantom{0}}$	$\overline{\phantom{0}}$	$\overline{\phantom{0}}$	3.36
	$[24-HSE]$		$\overline{\phantom{0}}$	-	-	3.86
	$[25-PBE]$		÷	-	-	0.773
	$[25-TB-mBJ]$	$\overline{\phantom{a}}$	-	-	-	3.354
	$[27-PBE]$	$\qquad \qquad -$	$\overline{\phantom{0}}$	-	-	0.57
	$[27-HSE]$	$\overline{\phantom{0}}$	$\overline{\phantom{0}}$	$\overline{\phantom{0}}$	$=$	3.86
ZnRh <sub>2</sub> O <sub>4</sub>	This work	2.17	2.06	0.99	2.00	1.14
	$[8-HSE]$	$\overline{\phantom{0}}$	$\overline{\phantom{0}}$	$\overline{\phantom{0}}$	$\overline{\phantom{0}}$	2.91
	$[9 - PBE]$	۳	$\overline{\phantom{0}}$	-	-	1.10
	$[13-Exp.]$		$\overline{\phantom{0}}$	-		2.74
	$[15]$		$\overline{\phantom{0}}$	-	$\overline{\phantom{0}}$	1.2
	$[18]$	$\overline{\phantom{a}}$	-	$\overline{\phantom{0}}$	$\overline{\phantom{0}}$	1.65
	$[24 - PBE]$	$\overline{\phantom{0}}$	$\overline{\phantom{0}}$	-	$\overline{\phantom{0}}$	0.89
	$[24-TB-mBJ]$	۰	-	-	$\overline{\phantom{0}}$	2.53
	$[24-HSE]$		$\overline{\phantom{0}}$	-		2.87
	$[25-PBE]$		-			0.897
	$[25-TB-mBJ]$	$\overline{\phantom{0}}$	$\overline{\phantom{0}}$	$\overline{\phantom{0}}$	$\overline{\phantom{0}}$	2.53
	$[27-PBE]$	$\overline{\phantom{a}}$	$\overline{\phantom{0}}$	$\overline{\phantom{0}}$	÷	0.80
	$[27-HSE]$			-	-	2.87
	$[29]$	$\overline{\phantom{a}}$	$\overline{\phantom{0}}$	$\overline{\phantom{0}}$	-	2.1
ZnIr <sub>2</sub> O <sub>4</sub>	This work	0.66	2.18	0.92	2.17	0.86
	$[8-HSE]$	$\qquad \qquad -$	$\overline{\phantom{a}}$	$\qquad \qquad -$	$\qquad \qquad -$	2.53
	$[11-B3LYP]$	$\overline{\phantom{a}}$	$\overline{\phantom{0}}$	$\overline{\phantom{0}}$	$\equiv$	3.06
	$[13-Exp.]$		$\overline{\phantom{0}}$	$\overline{\phantom{0}}$	-	2.97
	$[24 - PBE]$	-	$\overline{\phantom{0}}$	$\overline{\phantom{0}}$	÷	0.28
	$[24-TB-mBJ]$	۳	$\overline{\phantom{0}}$	-		2.30
	$[24-HSE]$	$\overline{\phantom{a}}$	$\overline{\phantom{0}}$	$\overline{\phantom{0}}$	$\equiv$	2.45
	$[25-PBE]$	$\overline{\phantom{a}}$	-	-	$\overline{\phantom{0}}$	0.416
	$[25-TB-mBJ]$	$\overline{\phantom{0}}$	$\overline{\phantom{0}}$	-	-	2.32
	$[27-PBE]$					0.48
	$[27-HSE]$			-		2.45
	$[28-LDA-mBJ]$	$\overline{\phantom{a}}$		$\equiv$		2.25

### $(C_{11} - C_{12}) > 0, C_{12} < B < C_{11}, (C_{11} + 2C_{12}) > 0, C_{44} > 0$  (1)

The elastic constant  $C_{11}$  represents the resistance in the x-direction against the linear compression. The elastic constants  $C_{11}$  of  $ZnCo<sub>2</sub>O<sub>4</sub>$ ,  $ZnRh<sub>2</sub>O<sub>4</sub>$  and  $ZnIr<sub>2</sub>O<sub>4</sub>$  compounds are larger than  $C<sub>12</sub>$  and  $C<sub>44</sub>$ , indicating that these compounds are less compressible along the x-direction. Unfortunately, no experimental or theoretical data are available for the elastic constants of these materials.

The B/G ratios for the three spinel oxides compounds were calculated. While materials with a B/G ratio of less than 1.75 exhibit brittle behaviour, materials with a B/G ratio greater than 1.75 exhibit ductile behaviour. The  $ZnCo_2O_4$ ,  $ZnRh_2O_4$  and  $ZnIr_2O_4$ compounds were shown to be ductile when their B/G ratio was greater than 1.75. Also, Cauchy pressure  $(C_{11}-C_{44})$  are classical criteria that determine whether the material is ductile or brittle. The positive or negative value of the Cauchy's pressure indicates that the material is ductile or brittle. In this study Cauchy-pressure was found to be positive in the press, so the compounds studied were confirmed to be ductile in nature.

In addition, Young's modulus (E) and Poisson's ratio  $(\nu)$  were calculated. The stiffness of the compounds was analysed by comparing the Young's modulus. The higher Young's modulus shows better stiffness, so ZnCo<sub>2</sub>O<sub>4</sub> is the hardest compound in this work. The Poisson ratio was calculated between the elastic parameters to determine the types of forces in the materials. According to the Pugh criteria [\[39\]](#page-8-16), if the Poisson ratio is around 0.1, the material has a covalent bond, if it is about 0.25, the material has an ionic bond [\[39\].](#page-8-16) In addition, the Poisson's ratio had a critical value of 0.26 indicating whether a material was ductile or brittle. The calculated  $\nu$  values indicated that all materials were ductile.

#### 3.4. Vibrational properties

The calculated full phonon dispersion curves, the total and projected density of states for spinel oxides  $\text{ZnM}_2\text{O}_4$  (M = Co, Rh, Ir) compounds in the spinel structure along the high-symmetry directions are shown in [Fig. 4](#page-6-0). According to the symmetry analysis spinel  $\text{ZnM}_2\text{O}_4$  (M = Co, Rh, Ir) compounds have 42 phonon modes, there are 39 optical modes distributed on the following symmetries at the Γ point.

<span id="page-5-0"></span>

Fig. 3. The total and partial density of states for cubic spinel  $\text{ZnM}_2\text{O}_4$  (M = Co, Rh, Ir) compounds.

 $\Gamma = A_{1g} + 2A_{2u} + E_g + 2E_u + T_{1g} + 4T_{1u} + 2T_{2u} + 3T_{2g}$  (2)

where  $T_{1u}$  modes are infrared active,  $A_{1g}$ ,  $E_g$  and  $T_{2g}$  modes are Raman actives and  $A_{2u}$ ,  $E_u$ ,  $T_{2u}$  and  $T_{1g}$  are silent (inactive). The phonon properties of ZnM<sub>2</sub>O<sub>4</sub> (M = Co, Rh, Ir) compounds are calculated within the generalized gradient approximation (GGA) in

<span id="page-6-0"></span>

Fig. 4. Full phonon spectra and projected phonon density of states of cubic spinel ZnM<sub>2</sub>O<sub>4</sub> (M = Co, Rh, Ir) compounds.

the spinel structure, with space group symmetry Fd-3 m (#227). From the phonon dispersion spectra, the band gap values were observed to be 3.85 THz and 1.84 THz for ZnIr<sub>2</sub>O<sub>4</sub> and ZnRh<sub>2</sub>O<sub>4</sub> compounds, respectively. ZnRh<sub>2</sub>O<sub>4</sub> exhibits similar phonon dispersion properties with the same space group (Fm-3 m, (225)) materials such as  $Rh_2TiAl$ ,  $Rh_2FeAl$  and  $MgXH_3$  (X = Fe, Co) [40–[42\].](#page-8-21) The calculated phonon dispersion curves of  $ZnM_2O_4$  (M = Co, Rh, Ir) compounds confirm that three compounds are dynamically

<span id="page-7-7"></span>

Fig. 5. Specific heats at constant volume versus temperature of cubic spinel  $\text{ZnM}_2\text{O}_4$  (M = Co, Rh, Ir) compounds.

stable in the spinel structure without any imaginary phonon frequencies. The  $\text{ZnM}_2\text{O}_4$  (M = Co, Rh, Ir) spinel compounds have not been compared with available data because there are no studies about their phonon frequencies.

#### 4. Conclusion

In this study, the structural, elastic, electronic and phonon properties of  $\text{ZnM}_2\text{O}_4$  (M = Co, Rh, Ir) in the spinel structure have been investigated using first principle density functional theory (DFT) within the generalized gradient approximation (GGA). The lattice constant  $(a_0)$ , the internal parameter (u), Bulk modulus and elastic constants for  $\text{ZnM}_2\text{O}_4$  (M = Co, Rh, Ir) are determined and evaluated. It can be seen that the  $ZnCo<sub>2</sub>O<sub>4</sub>$ ,  $ZnRh<sub>2</sub>O<sub>4</sub>$  and  $ZnIr<sub>2</sub>O<sub>4</sub>$  compounds show ductile manner and meets the well-known mechanical stability criteria for crystals. From the calculated electronic band spectra, we obtained indirect band gap values of 1.25 eV, 1.14 eV and 0.86 eV for  $ZnCo<sub>2</sub>O<sub>4</sub>$ ,  $ZnRh<sub>2</sub>O<sub>4</sub>$  and  $ZnIr<sub>2</sub>O<sub>4</sub>$  spinels, respectively. The phonon dispersion curves of  $ZnCo<sub>2</sub>O<sub>4</sub>$ , ZnRh2O4 and ZnIr2O4 spinel compounds along all symmetry directions in the Brillouin zone have been investigated for the first time in this work. From phonon dispersion curves, there is a gap between optical-optic phonon modes, except for the  $ZnCo<sub>2</sub>O<sub>4</sub>$  spinel compound. This gap is caused by the mass difference between the atoms. It also shows that these alloys are dynamically stable because the phonon frequencies are not imaginary. [Fig. 5](#page-7-7) show the calculated the specific heat capacity  $(C_v)$  at a constant volume of spinel oxides  $\text{ZnM}_2\text{O}_4$  (M = Co, Rh, Ir) compounds. The C<sub>v</sub> values of these compounds increase rapidly in the 250 K range before starting the saturation point. At low temperatures, acoustic modes are important where optical modes take over at elevated temperatures due to electrostatic interactions. At a high-temperature, the calculated specific heat capacity  $C_v$  of these compounds are nearly to the Dulong-Petit limit [\[42,43\]](#page-8-22). Finally, using the computed phonon density of state, the specific heats  $(C_v)$  at constant volume has been determined using by the quasi harmonic approximation (QHA).

#### References

- <span id="page-7-0"></span>[1] [J. Robertson, B. Falabretti, Electronic structure of transparent conducting oxides, in: D.S. Ginley \(Ed.\), Handbook of Transparent Conductors, Springer, US, 2011,](http://refhub.elsevier.com/S0577-9073(18)31146-8/sbref0001) [pp. 27](http://refhub.elsevier.com/S0577-9073(18)31146-8/sbref0001)–50.
- [2] [A. Stadler, Transparent conducting oxides-An up-to-date overview, Materials 5 \(4\) \(2012\) 661](http://refhub.elsevier.com/S0577-9073(18)31146-8/sbref0002)–683.
- <span id="page-7-1"></span>[3] [X. Tao, V. Koncar, C. Dufour, A coating of polyethylene terephthalate nonwoven fabrics with indium zinc oxide solutions at low temperature by sol-gel process, J.](http://refhub.elsevier.com/S0577-9073(18)31146-8/sbref0003) [Textiles Eng. 79 \(2013\) 2](http://refhub.elsevier.com/S0577-9073(18)31146-8/sbref0003)–6.
- [4] [E. Savarimuthu, K.C. Lalithambika, A.M.E. Raj, L.C. Nehru, S. Ramamurthy, A. Thayumanavan, C. Sanjeeviraja, M. Jayachandran, Synthesis and materials](http://refhub.elsevier.com/S0577-9073(18)31146-8/sbref0004) properties of transparent conducting In2O3 films prepared by sol–[gel-spin coating technique, J. Phys. Chem. Solids 68 \(7\) \(2007\) 1380](http://refhub.elsevier.com/S0577-9073(18)31146-8/sbref0004)–1389.

[5] Z.H. Li, D.Y. Ren, Preparation of ITO transparent conductive fi[lm by sol-gel method, Trans. Nonferrous Metals Soc. China 16 \(6\) \(2006\) 1358](http://refhub.elsevier.com/S0577-9073(18)31146-8/sbref0005)–1361.

- [6] [M.A. Aegerter, N. Al-Dahoudi, Wet-chemical processing of transparent and antiglare conducting ITO coating on plastic substrates, J. Sol-Gel Sci. Technol. 27 \(1\)](http://refhub.elsevier.com/S0577-9073(18)31146-8/sbref0006) [\(2003\) 81](http://refhub.elsevier.com/S0577-9073(18)31146-8/sbref0006)–89.
- [7] A. Candan, G Uğur, First-principles study of structural, electronic, elastic and phonon properties of AB<sub>2</sub>O<sub>4</sub> (A = Ge, Si; B = Mg, Zn, Cd) spinel oxides, Mod. Phys. [Lett. B 30 \(03\) \(2016\) 1650002.](http://refhub.elsevier.com/S0577-9073(18)31146-8/sbref0007)
- <span id="page-7-2"></span>[8] M.N. Amini, H. Dixit, R. Saniz, D. Lamoen, B Partoens, The origin of p-type conductivity in ZnM<sub>2</sub>O<sub>4</sub> (M = Co, Rh, Ir) spinels, Phys. Chem. Chem. Phys. 16 (6) [\(2014\) 2588](http://refhub.elsevier.com/S0577-9073(18)31146-8/sbref0008)–2596.
- <span id="page-7-3"></span>[9] O. Volnianska, P. Boguslawski, Calculated properties of point native defects and p-type conductivity of ZnRh<sub>2</sub>O<sub>4</sub>, J. Appl. Phys. 114 (3) (2013) 033711.
- <span id="page-7-6"></span>[10] S. Samanta, S.M. Saini, Full-potential study of the electronic and optical properties of the transparent oxide ZnCo<sub>2</sub>O<sub>4</sub> by use of PBE and TB-mBJ potentials, J. [Electron. Mater. 43 \(9\) \(2014\) 3659](http://refhub.elsevier.com/S0577-9073(18)31146-8/sbref0010)–3665.
- <span id="page-7-4"></span>[11] D.M. Ramo, P.D. Bristowe, The effect of defects and disorder on the electronic properties of ZnIr<sub>2</sub>O<sub>4</sub>, J. Chem. Phys. 141 (8) (2014) 084704.
- [12] O. Volnianska, P. Boguslawski, GGA+ U study of native point defects in ZnRh<sub>2</sub>O<sub>4</sub>, J. Phys. D Appl. Phys. 47 (46) (2014) 465101.
- <span id="page-7-5"></span>[13] [M. Dekkers, G. Rijnders, D.H. Blank, ZnIrO, a p-type transparent oxide semiconductor in the class of spinel zinc-d6-transition metal oxide, Appl. Phys. Lett. 90 \(2\)](http://refhub.elsevier.com/S0577-9073(18)31146-8/sbref0013) [\(2007\) 21903 -21903.](http://refhub.elsevier.com/S0577-9073(18)31146-8/sbref0013)

- <span id="page-8-0"></span>[14] Y. Sharma, N. Sharma, G.V. Subba Rao, B.V.R. Chowdari, Nanophase ZnCo<sub>2</sub>O<sub>4</sub> as a High Performance Anode material for Li-Ion batteries, Adv. Funct. Mater. 17 [\(15\) \(2007\) 2855](http://refhub.elsevier.com/S0577-9073(18)31146-8/sbref0014)–2861.
- <span id="page-8-1"></span>[15] D.J. Singh, R.C. Rai, J.L. Musfeldt, S. Auluck, N. Singh, P. Khalifah, D.G. Mandrus, Optical properties and electronic structure of spinel ZnRh<sub>2O4</sub>, Chem. Mater. 18 [\(11\) \(2006\) 2696](http://refhub.elsevier.com/S0577-9073(18)31146-8/sbref0015)–2700.
- <span id="page-8-17"></span>[16] [S. Kim, J.A. Cianfrone, P. Sadik, K.W. Kim, M. Ivill, D.P. Norton, Room temperature deposited oxide pn junction using p-type zinc-cobalt-oxide, J. Appl. Phys.](http://refhub.elsevier.com/S0577-9073(18)31146-8/sbref0016) [107 \(10\) \(2010\) 103538.](http://refhub.elsevier.com/S0577-9073(18)31146-8/sbref0016)
- B. Chi, J. Li, X. Yang, H. Lin, N. Wang, Electrophoretic deposition of ZnCo<sub>2</sub>O<sub>4</sub> spinel and its electrocatalytic properties for oxygen evolution reaction, [Electrochim. Acta 50 \(10\) \(2005\) 2059](http://refhub.elsevier.com/S0577-9073(18)31146-8/sbref0017)–2064.
- <span id="page-8-18"></span>[18] N. Mansourian-Hadavi, S. Wansom, N.H. Perry, A.R. Nagaraja, T.O. Mason, L.H. Ye, A.J. Freeman, Transport and band structure studies of crystalline ZnRh<sub>2O4</sub>, [Phys. Rev. B 81 \(7\) \(2010\) 075112.](http://refhub.elsevier.com/S0577-9073(18)31146-8/sbref0018)
- [19] H.J. Kim, I.C. Song, J.H. Sim, H. Kim, D. Kim, Y.E. Ihm, & W.K. Choo, Structural and transport properties of cubic spinel ZnCo<sub>2</sub>O<sub>4</sub> thin films grown by reactive magnetron sputtering, Solid State Commun. 129(10) (2004) 627–630.
- <span id="page-8-11"></span>[20] [G.Y. Zhang, B. Guo, J. Chen, MCoO \(M = Ni, Cu, Zn\) nanotubes: template synthesis and application in gas sensors, Sens. Actuators B 114 \(1\) \(2006\) 402](http://refhub.elsevier.com/S0577-9073(18)31146-8/sbref0019)-409.
- <span id="page-8-10"></span>[21] X. Wei, D. Chen, W. Tang, Preparation and characterization of the spinel oxide ZnCo<sub>2</sub>O<sub>4</sub> obtained by sol-gel method, Mater. Chem. Phys. 103 (1) (2007) 54–58.
- [22] [X. Niu, W. Du, W. Du, Preparation and gas sensing properties of ZnM2](http://refhub.elsevier.com/S0577-9073(18)31146-8/sbref0021) O4 (M= Fe, Co, Cr), Sens. Actuators B 99 (2) (2004) 405–409.
- [23] H.J. Park, J. Kim, N.J. Choi, H. Song, D.S. Lee, Nonstoichiometric Co-rich ZnCo<sub>2</sub>O<sub>4</sub> hollow nanospheres for high performance formaldehyde detection at ppb [levels, ACS Appl. Mater. Interf. 8 \(5\) \(2016\) 3233](http://refhub.elsevier.com/S0577-9073(18)31146-8/sbref0022)–3240.
- <span id="page-8-12"></span>[24] [H. Dixit, R. Saniz, S. Cottenier, D. Lamoen, B. Partoens, Electronic structure of transparent oxides with the Tran](http://refhub.elsevier.com/S0577-9073(18)31146-8/sbref0023)–Blaha modified Becke–Johnson potential, J. [Phys. Condens. Matter 24 \(20\) \(2012\) 205503.](http://refhub.elsevier.com/S0577-9073(18)31146-8/sbref0023)
- <span id="page-8-2"></span>[25] S. Samanta, Study of systematic trends in electronic and optical properties within ZnM<sub>2</sub>O<sub>4</sub> (M = Co, Rh, Ir) family by FPLAPW method with PBE and TB-mBJ [potentials, Opt. Mater. 45 \(2015\) 141](http://refhub.elsevier.com/S0577-9073(18)31146-8/sbref0024)–147.
- <span id="page-8-13"></span>[26] [F.K. Lotgering, On the ferrimagnetism of some sulphides and oxides, Philips Res. Rep. 11 \(1956\) 190.](http://refhub.elsevier.com/S0577-9073(18)31146-8/sbref0025)
- <span id="page-8-14"></span>[27] D.O. Scanlon, G.W. Watson, Band gap anomalies of the  $ZnM_2IIIO_4$  (MIII = Co, Rh, Ir) spinels, PCCP 13 (20) (2011) 9667–9675.
- <span id="page-8-20"></span>[28] N. Singh, U. Schwingenschlögl, ZnIrO: an effi[cient photocatalyst with Rashba splitting, Europhys. Lett. 104 \(3\) \(2013\) 37002.](http://refhub.elsevier.com/S0577-9073(18)31146-8/sbref0027)
- <span id="page-8-19"></span>[29] [H. Mizoguchi, M. Hirano, S. Fujitsu, T. Takeuchi, K. Ueda, H. Hosono, ZnRhO: a p-type semiconducting oxide with a valence band composed of a low spin state of](http://refhub.elsevier.com/S0577-9073(18)31146-8/sbref0028) Rh 3+ in a 4d 6 confi[guration, Appl. Phys. Lett. 80 \(7\) \(2002\) 1207](http://refhub.elsevier.com/S0577-9073(18)31146-8/sbref0028)–1209.
- <span id="page-8-3"></span>[30] [G. Kresse, J. Hafner, Ab. initio molecular dynamics for liquid metals, Phys. Rev. B 47 \(1993\) 558.](http://refhub.elsevier.com/S0577-9073(18)31146-8/sbref0029)
- [31] G. Kresse, J. Furthmüller, Effi[cient iterative schemes for ab initio total-energy calculations using a plane-wave basis set, Phys. Rev. B 54 \(1996\) 11169](http://refhub.elsevier.com/S0577-9073(18)31146-8/sbref0030)–11186.
- <span id="page-8-4"></span>[32] [P.E. Blöchl, Projector augmented-wave method, Phys. Rev. B 50 \(1994\) 17953](http://refhub.elsevier.com/S0577-9073(18)31146-8/sbref0031)–17979.
- <span id="page-8-5"></span>[33] [J.P. Perdew, K. Burke, M. Ernzerhof, Generalized gradient approximation made simple, Phys. Rev. Lett. 77 \(1996\) 3865](http://refhub.elsevier.com/S0577-9073(18)31146-8/sbref0032)–3868.
- <span id="page-8-6"></span>[34] [H.J. Monkhorst, J.D. Pack, Special points for Brillouin-zone integrations, Phys. Rev. B 13 \(1976\) 5188.](http://refhub.elsevier.com/S0577-9073(18)31146-8/sbref0033)
- <span id="page-8-7"></span>[35] [M. Methfessel, A.T. Paxton, High-precision sampling for Brillouin-zone integration in metals, Phys. Rev. B 40 \(1989\) 3616.](http://refhub.elsevier.com/S0577-9073(18)31146-8/sbref0034)
- <span id="page-8-9"></span><span id="page-8-8"></span>[36] [Y.L Page, P. Saxe, Symmetry-general least-squares extraction of elastic coe](http://refhub.elsevier.com/S0577-9073(18)31146-8/sbref0035)fficients from ab initio total energy calculations, Phys. Rev. B 63 (2001) 174103. [37] A. Candan, G. Uğur, Z. Charifi, H. Baaziz, M.R. Ellialtıoğ[lu, Electronic structure and vibrational properties in cobalt-based full-Heusler compounds: a](http://refhub.elsevier.com/S0577-9073(18)31146-8/sbref0036) first principle study of  $Co<sub>2</sub>MnX$  (X = Si, Ge, Al, Ga), J. Alloys Compd. 560 (2013) 215–222.
- <span id="page-8-15"></span>[38] [M. Born, K. Huang, Dynamical Theory of Crystal Lattices, Clarendon Press, Oxford, 1954, p. 420.](http://refhub.elsevier.com/S0577-9073(18)31146-8/sbref0037)
- <span id="page-8-16"></span>[39] [S.F. Pugh, XCII. Relation between the elastic moduli and the plastic properties of polycrystalline pure metals, Philos. Mag. J. Sci. 45 \(1954\) 823.](http://refhub.elsevier.com/S0577-9073(18)31146-8/sbref0038)
- <span id="page-8-21"></span>[40] [S. Al, N. Arikan, A. Iyigör, Investigations of structural, elastic, electronic and thermodynamic properties of X2TiAl alloys: a computational study, Zeitschrift für](http://refhub.elsevier.com/S0577-9073(18)31146-8/sbref0039) [Naturforschung A 73 \(9\) \(2018\) 859](http://refhub.elsevier.com/S0577-9073(18)31146-8/sbref0039)–867.
- [41] S. Al, et al., Lattice dynamic properties of  $Rh_2XAI$  (X = Fe and Y) alloys, Physica B 531 (2018) 16–20.
- <span id="page-8-22"></span>[42] G.D. Yıldı[z, et al., Computational investigations of mechanic, electronic and lattice dynamic properties of yttrium based compounds, Int. J. Modern Phys. B 32](http://refhub.elsevier.com/S0577-9073(18)31146-8/sbref0041) [\(20\) \(2018\) 1850214.](http://refhub.elsevier.com/S0577-9073(18)31146-8/sbref0041)
- [43] [A.T. Petit, Recherches sur quelques points importans de la theorie de la Chaleur, Ann. Chim. Phys. 10 \(1819\) 395.](http://refhub.elsevier.com/S0577-9073(18)31146-8/sbref0042)